

MARILING SCHEME: GATUNGU EVALUATION EXAM

1. Volume = $150 - 80 = 70 \text{ cm}^3 \checkmark$

$$\rho = \frac{M}{V} \checkmark$$

$$\rho = \frac{567}{70} \text{ or}$$

$$= 8.10 \text{ g/cm}^3 \checkmark$$

2. (i) - Weight of solid in air (Real weight) \checkmark

- Weight of solid in the fluid (Apparent weight) \checkmark

(ii) Upthrust = Real weight - Apparent weight \checkmark

3. The object will continue to move with constant velocity along a straight line. \checkmark

4. Presence of impurities \checkmark

- Increase in pressure \checkmark

5. Flask contracts first \checkmark

Water contracts more than flask.

6. Blowing increases the velocity of air inside the tube which leads to decrease in pressure. \checkmark

The atmospheric pressure pushes the walls of the tube.

7) $F_1 d_1 = F_2 d_2$ or sum of clockwise moment = sum of anticlockwise moments \checkmark

$$0.5 \times T + 4 \times 0.35 = 0.4 \times 8 \checkmark$$

$$0.5T + 1.4 = 3.2$$

$$T = \frac{1.8}{0.5} = 3.6 \checkmark$$

8. (a) $F = 20\text{N}$

$l = 0.5\text{cm}$

$k = \frac{F}{l} \checkmark$

$k = \frac{20\text{N}}{0.5\text{cm}}$

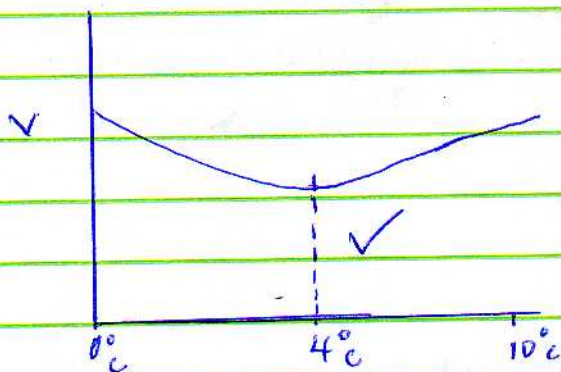
$= 40\text{N/cm} \checkmark$ or 0.4N/m

(b) $F = ke$

$= 40\text{N/cm} \times 0.86\text{cm}$

$= 34.4\text{N} \checkmark$

9.



10. (a) Melting or solid changes to liquid \checkmark

(b) Vapourisation (liquid changes to gas).

(ii) Prevent mercury from flowing back into bulb. \checkmark

12. Diffusion \checkmark

13. The position of the centre of gravity (C.O.G) does not change even with a displacement. \checkmark or The C.O.G is at its lowest point.

14(a) Gas that obeys all gas laws (perfectly) ✓

by changing pressure slowly ✓ or allowing gas to go back to original temperature after the change.

(ii) $k =$ The slope ✓

$(2.0 \times 10^6, 1.65 \times 10^5)$ (reading from the graph)
and

$(4.0 \times 10^6, 3.3 \times 10^5)$

$$\frac{(3.3 - 1.65) \times 10^5}{(4.0 - 2.0) \times 10^6} = \frac{1.65 \times 10^5}{2.0 \times 10^6} \checkmark$$
$$= 0.0825 \text{ Nm} \checkmark$$

(iii) Work done on the gas.

(iv) - use dry gas or ✓
make very small changes in temperature.

c) $\frac{V_1}{T_1} = \frac{V_2}{T_2} \checkmark$

or

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \text{ where } P_1 = P_2$$

$$V_2 = V_1 T_2$$