

FORM 2 TERM 1 OPENER

CHEMISTRY

NAME..... ADM NO.....

CLASS..... SIGN.....

DATE.....

TIME:2 HOURS

INSTRUCTIONS

Answer all questions in the spaces provided

1. From the following list of compounds; zinc oxide, solid carbon (IV) oxide, sodium carbonate, nitric (V) acid, iron (III) chloride;

(i) Identify two that sublime. (2 mks)

(ii) Identify a pair that react to form salt and water only. (2 mks)

(iii) Write a word equation for the reaction between sodium carbonate and nitric (V) acid. (1 mk)

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2. In an experiment to investigate the percentage of oxygen in air, 200 cm³ of air was passed over heated copper turning repeatedly until a constant volume of air remained. 160 cm³ of air remained at the end of the experiment.

(i) Name one gas remaining in the 160 cm³ of air. (1 mk)

(ii) Determine the percentage of air used up during the experiment. (2 mks)

(iii) What observation would be made during the experiment. (1 mk)

(iv) Write a word equation for the reaction between copper and oxygen. (1 mk)

3. Name the best method to use to separate the following mixtures. (4 mks)

(i) Common salt and water.

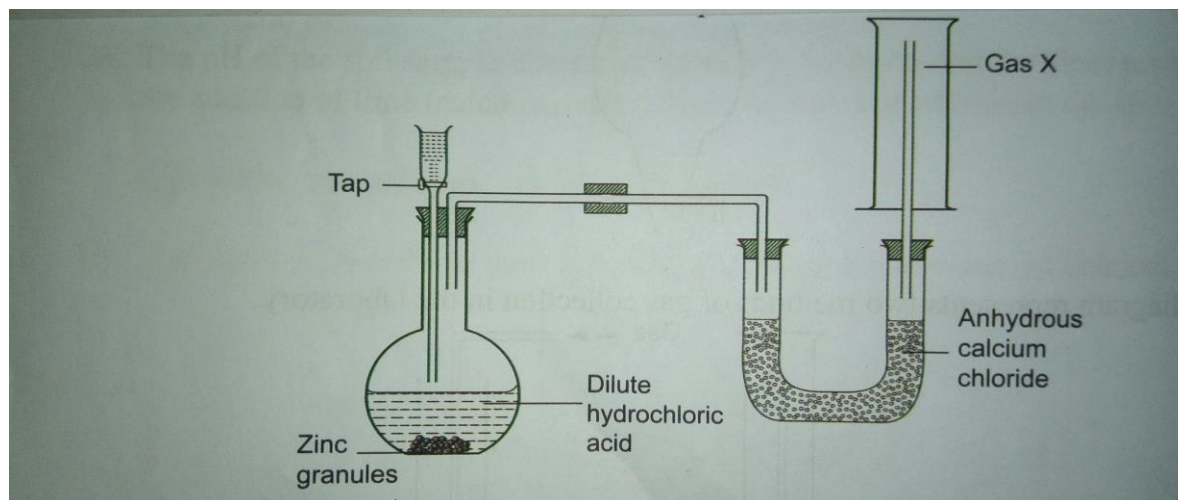
(ii) Coloured dyes in ethanol.

(iii) Ammonium chloride and sodium chloride.

(iv) Ethanol and water.

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4. Study the set up below and answer the questions that follow.



(a) Identify gas X. (1 mk)

(b) Name the method used to collect gas X and give the property of the gas that enables it to be collected using the method. (2 mks)

(c) Write a word equation for the reaction liberating gas X. (1 mk)

(d) Why is not advisable to use the following in this method of preparing hydrogen;

(i) Calcium and dilute sulphuric acid (1 mk)

(ii) Zinc and Nitric (V) acid. (1mk)

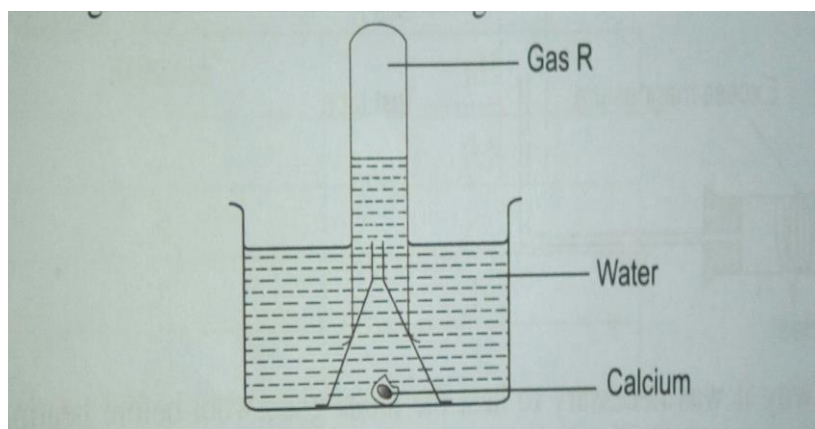
(e) What is the purpose of anhydrous calcium chloride in the U-tube? (1mk)

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(f) Name another compound that could serve the same purpose as anhydrous calcium chloride. (1 mk)

5. When magnesium is burnt in air, it reacts with oxygen and nitrogen gas giving a white ash. Write two equations for the two reactions that take place. (2 mks)

6. The diagram below was used to investigate the action of water on calcium metal.



(a) Identify gas R. (1 mk)

(b) The remaining solution in the beaker changes red litmus paper to blue. What does this suggest about the resulting solution. (1 mk)

(c) Write a word equation for the reaction. (1 mk)

(d) What would be observed if Carbon(IV) oxide is bubbled through the solution. (1 mk)

7. a) Name the particles that are found in an atom. (3 mks)

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b) Atoms are said to be electrically neutral. Explain. (2 mks)

c) Element Q has 18 neutrons and a mass number of 34.

i) How many protons are present in element Q? (1 mk)

(ii) Write the electron arrangement of element Q. (1 mk)

(iii) To which period and group does Q belong? Explain your answer. (2 mks)

8. (a) In an experiment to investigate what happens when steam is passed over heated magnesium, wet sand or cotton wool soaked in water is heated first before heating magnesium. Why is it necessary? (1 mk)

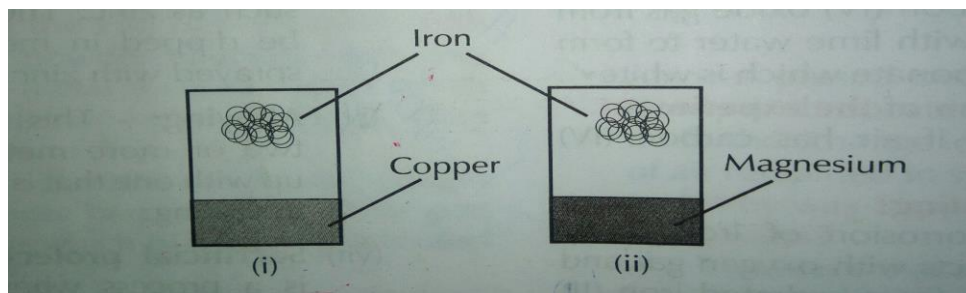
(b) Write a word equation for the reaction between magnesium and steam. (1 mk)

(c) Write a word equation for the reaction between iron and steam. (1 mk)

9. (a) What is the chemical name of rust? (1 mk)

(b) A form one student in attempt to stop rusting put copper and magnesium in contact with iron as shown.

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(i) The set ups were left outside for two weeks. Where did rusting occur? Explain. (2 mks)

(ii) Name the method used by the student to prevent rusting. (1 mk)

(iii) How does painting prevent rusting? (1 mk)

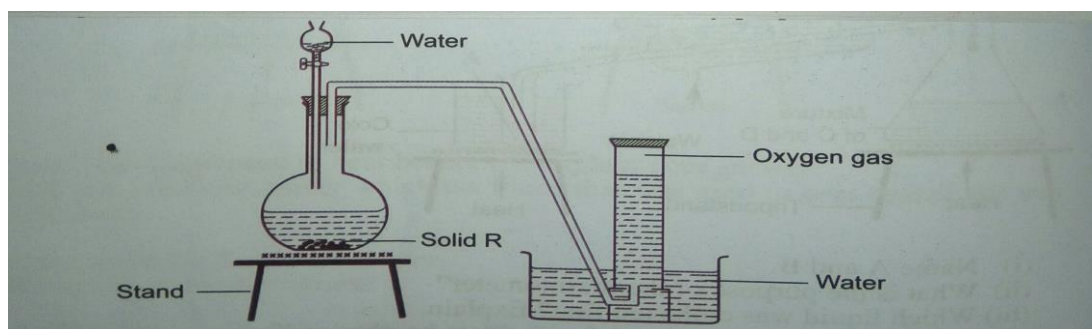
10. State the use of the following apparatus as used in the laboratory.

(a) Pipe-clay triangle (1 mk)

(b) Desiccator (1 mk)

(c) Wire gauze (1 mk)

11. The diagram below represents a set up for the laboratory preparation of oxygen gas.



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(i) Name solid R.(1 mk)

(ii)Write a word equation for the reaction in the flask.(1 mk)

(iii)How can one test for oxygen gas?(2 mks)

(iv)Give two commercial uses of oxygen gas.(2 mks)

12. A student was supplied with a colourless liquid suspected to be water.

(i) Describe one test that could be carried out to show that the liquid was water.(2 mks)

(ii)How could it have been shown that the liquid was pure water?(1 mk)

13. When a student was stung by a stinging nettle plant,a teacher applied an aqueous solution of ammonia to the affected area of the skin and the student was relieved of pain.Explain.(2 mks)

14. A mixture of magnesium powder and lead (II)oxide will react vigorously when heated but no reaction occurs when a mixture of magnesium oxide and lead powder are heated.

(a)Explain the observations above.(2 mks)

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(b) Write a word equation for the reaction between magnesium and lead(II) oxide. (1 mk)

(c) From (b) above, identify the; (i) oxidised substance. (1 mk)

(ii) oxidizing agent. (1 mk)

(iii) what name is given to such a reaction? (1 mk)

15. Classify the following as either physical or chemical changes. (5 mks)

a) Freezing of beer

(b) Rusting of iron

(c) Heating of glass until it melts

(d) Burning a candle.

(e) Heating copper(II) nitrate

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All the Best

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